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Key indicators Single-crystal X-ray study

T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.059 wR factor = 0.170 Data-to-parameter ratio = 17.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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1'-Methyl-4'-(3,4,5-trimethoxyphenyl)dispiro-[indene-2,3'-pyrrolidine-2',3"-indole]-1,2",3(1"H)-trione

The pyrrolidine ring of the title compound, $C_{29}H_{26}N_2O_6$, adopts a twist conformation. The structure is stabilized by weak intramolecular $C-H\cdots O$ interactions as well as by intermolecular $C-H\cdots O$, $N-H\cdots O$ and $N-H\cdots N$ interactions. Received 14 January 2004 Accepted 3 March 2004 Online 20 March 2004

Comment

Several unusual amino acids which contain the pyrrolidine motif have been investigated by Galeazzi *et al.* (1999). Furthermore, pyrrolidine compounds are capable of exhibiting antimicrobial and antifungal activity (Amal Raj *et al.*, 2003). As part of our ongoing studies of spiro-pyrrolidines, the crystal structure analysis of the title compound, (I), has been carried out and the results are presented here.

The C-C bond lengths (Table 1) in pyrrolidine ring B are longer than normal, perhaps due to the three bulky substituents. A similar effect has been found in a related structure reported by Jeyabharathi *et al.* (2001).



The geometry of the oxindole moiety is normal and agrees very well with that in other publications (Govind *et al.*, 2003; Govindasamy *et al.*, 1999; Sethu Sankar *et al.*, 2002). The Cambridge Structural Database (CSD) November 2003 Release (Allen, 2002) contains 20 examples of this geometry.

The bond lengths and angles of the trimethoxyphenyl and indandione moieties are perfectly normal, as evidenced by 252 and 34 examples, respectively, in the CSD.

Ring A is in an envelope conformation with lowest asymmetry parameter (Nardelli, 1983) $\Delta C_s(C4) = 0.009$ (1); atom C4 deviates by 0.046 (2) Å from the least-squares plane passing through the remaining four atoms (N1, C2, C5 and C10) of the ring. Pyrrolidine ring B adopts a twist conformation, with puckering parameters (Cremer & Pople, 1975) $q_2 = 0.441$ (2) Å and $\varphi = 17.1$ (2)°, and asymmetry parameter $\Delta C_2(C13) = 0.008$ (1). Ring C is also in a twist conformation, with puckering parameters $q_2 = 0.096$ (2) Å and $\varphi = -170.6$ (2)°, and asymmetry parameter $\Delta C_2(C29) = 0.008$ (1).



Figure 1

View of (I), with the atom-numbering scheme and displacement ellipsoids drawn at the 50% probability level.



Figure 2

A view of the packing of the molecules in the unit cell; dashed lines indicate hydrogen bonds.

In addition to van der Waals interactions, the crystal structure is stabilized by weak $C-H\cdots O$ intramolecular interactions, and also by $C-H\cdots O$, $N-H\cdots O$ and $N-H\cdots N$ intermolecular interactions.

Experimental

A mixture of 3,4,5-trimethoxybenzylidine (1 mmol), 1,3-indandione (1 mmol) and sarcosine (1 mmol) was refluxed in aqueous methanol. The resulting crude product was filtered off and recrystallized from methanol.

Crystal data

C ₂₉ H ₂₆ N ₂ O ₆	$D_x = 1.295 \text{ Mg m}^{-3}$
$M_r = 498.52$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 3201
a = 10.3884 (7) Å	reflections
b = 23.2621 (14) Å	$\theta = 2.4 - 23.0^{\circ}$
c = 11.1685 (7) Å	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 108.633 \ (1)^{\circ}$	T = 293 (2) K
V = 2557.5 (3) Å ³	Block, yellow
Z = 4	$0.20 \times 0.19 \times 0.19 \text{ mm}$

Data collection

1

Bruker SMART APEX CCD area-	5915 independent reflections
detector diffractometer	3919 reflections with $I > 2\sigma(I)$
v scans	$R_{int} = 0.035$
Absorption correction: multi-scan	$\theta_{max} = 28.0^{\circ}$
(<i>SADABS</i> ; Sheldrick, 2001)	$h = -13 \rightarrow 13$
$T_{min} = 0.982, T_{max} = 0.983$	$k = -30 \rightarrow 26$
15 973 measured reflections	$l = -14 \rightarrow 14$
Refinement	
Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.059$	$w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$
$wR(F^2) = 0.170$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.00	$(\Delta/\sigma)_{max} < 0.001$
5915 reflections	$\Delta\rho_{max} = 0.34 \text{ e } \text{Å}^{-3}$
334 parameters	$\Delta\rho_{min} = -0.20 \text{ e } \text{Å}^{-3}$
Fable 1	

Selected geometric parameters (Å, °).

C4-N14	1.452 (2)	C12-C13	1.537 (3)
C4-C11	1.556 (2)	C13-N14	1.475 (2)
C11-C12	1.563 (2)		
N14-C4-C11	101.0 (1)	N14-C13-C12	106.3 (1)
C4-C11-C12	101.0(1)	C4-N14-C13	108.5 (1)
C13-C12-C11	102.8 (1)		

 Table 2

 Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
C6-H6···O23	0.93	2.51	3.150 (3)	126
$C12 - H12 \cdot \cdot \cdot O3$	0.98	2.41	3.010 (2)	119
C21-H21···O23	0.93	2.58	3.207 (3)	125
C35−H35C···O32	0.96	2.49	3.050 (4)	117
$N1 - H1 \cdot \cdot \cdot O23^{i}$	0.86	2.45	3.125 (2)	136
$N1 - H1 \cdots N14^{i}$	0.86	2.27	3.006 (2)	143
C8−H8···O3 ⁱⁱ	0.93	2.47	3.331 (3)	155
$C13 - H13A \cdots O3^{iii}$	0.97	2.45	3.400 (2)	167
$C21 - H21 \cdots O3^{iii}$	0.93	2.58	3.336 (3)	139

Symmetry codes: (i) $\frac{1}{2} + x$, $\frac{1}{2} - y$, $\frac{1}{2} + z$; (ii) $x - \frac{1}{2}$, $\frac{1}{2} - y$, $\frac{1}{2} + z$; (iii) $x - \frac{1}{2}$, $\frac{1}{2} - y$, $z - \frac{1}{2}$.

All H atoms were positioned geometrically and allowed to ride on their parent atoms, with C-H = 0.93–0.98 Å and N-H = 0.86 Å, and with $U_{iso}(H) = 1.5U_{eq}(C)$ for methyl H atoms, $1.2U_{eq}(C)$ for other C-bound H atoms and $U_{iso}(H) = 1.2U_{eq}(N)$ for the indole H atom.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ZORTEP* (Zsolnai, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL*97 and *PARST* (Nardelli, 1995).

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References

- Allen, F. H. (2002) Acta Cryst B58, 380-388.
- Amal Raj, A., Raghunathan, R., Sridevikumari, M. R. & Raman, N. (2003). Bioorg. Med. Chem. 11, 407–419.

Bruker (2001). SAINT (Version 6.28a) and SMART (Version 5.625). Bruker AXS Inc., Madison, Wisconsin, USA.

Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.

- Galeazzi, R., Geremia, S., Mobbilli, G. & Orena, M. (1999). Tetrahedron: Asymmetry, 10, 587-605.
- Govind, M. M., Govindaraj, J., Rajakannan, V., Velmurugan, D., Kim, M.-J., Srinivasan, P. C. & Kannadasan, S. (2003). Acta Cryst. E59, o177– 0179.
- Govindasamy, L., Velmurugan, D., Shanmuga Sundara Raj, S. & Fun, H.-K. (1999). Acta Cryst. C55, 1315–1317.
- Jeyabharathi, A., Ponnuswamy, M. N., Amal Raj, A., Raghunathan, R., Razak, I. A., Usman, A., Chantrapromma, S. & Fun, H.-K. (2001). Acta Cryst. E57, 0901–0903.
- Nardelli, M. (1983). Acta Cryst. C39, 1141-1142.
- Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
- Sethu Sankar, K., Kannadasan, S., Velmurugan, D., Srinivasan, P. C., Shanmuga Sundara Raj, S. & Fun, H.-K. (2002). *Acta Cryst.* C58, o277–o279.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
 Sheldrick, G. M. (2001) SADARS, Varian 203, University of Göttingen
- Sheldrick, G. M. (2001). SADABS. Version 2.03. University of Göttingen, Germany.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Zsolnai. L. (1997). ZORTEP. University of Heidelberg, Germany.